

Synthesis of Tetracyclic Quinazolinones Using a Visible-Light-**Promoted Radical Cascade Approach**

Yue-Yue Han, Heng Jiang, Ruzhi Wang, and Shouyun Yu*

State Key Laboratory of Analytical Chemistry for Life Science, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210023, China

Supporting Information

ABSTRACT: A practical approach for the synthesis of tetracyclic pyrroloquinazolines using photoredox strategy has been developed. The visible-light-promoted intramolecular single-electron-transfer process between photocatalyst and N-(2-iodobenzyl)-N-acylcyanamides is considered to be involved in this transformation. Targeted pyrrologuinazoline derivatives (15 examples) are presented in good isolated yields (30%-88%).

tracyclic pyrroloquinazoline occurs as a key structural entity in dozens of natural products, such as luotonins A and E, mackinazolinone, rutaecarpine, and deoxyvasicinone (Scheme 1). These types of natural products disclose a wide

Scheme 1. Quinazolinone-Containing Biologically Active **Natural Products**

range of biological activities including the specific cytotoxicity and potent inhibition to proteases, which would be applicable in the discovery of pharmaceutical candidates.² Therefore, the synthesis of heterocyclic compounds bearing the tetracyclic pyrroloquinazoline scaffold has drawn intensive attention to synthetic chemists. Previously, the construction of tetracyclic pyrroloquinazoline scaffold commonly focused on the Pdcatalyzed cyclizations.³ For example, Li and co-workers reported a Pd-catalyzed approach to prepare tetracyclic pyrroloquinazolines from 2-bromo-N-(2-iodobenzyl)benzamides, which underwent cyanation/N-addition/N-arylation reaction sequence in a two-stage and one-pot manner (Scheme 2a).3a The Wu group also reported a decent method to prepare tetracyclic pyrroloquinazolines through palladiumcatalyzed carbonylation of 2-bromobenzylamines with 2-bromoanilines (Scheme 2b).^{3b} In addition, tetracyclic pyrroloquinazolines could also be prepared through radical

Scheme 2. Synthesis of Tetracyclic Quinazolinones

a) Pd-catalyzed sequential cyanation/N-addition/N-arylation

b) Palladium-catalyzed carbonylation

c) AIBN/Bu₃SnH-mediated radical cascade cyclization

d) Visible-light-promoted radical cascade cyclization: this work

approach. 4,5 Malacria and co-workers described the AIBN/n-Bu₃SnH-mediated radical cascade cyclization of N-(2-iodobenzyl)-N-acylcyanamides to prepare pyrroloquinazoline-type polycyclic compounds (Scheme 2c). ^{4a} A concise total synthesis of luotonin A was achieved using this strategy. The aforementioned methods are normally conducted at elevated temperature or proceed in the presence of toxic reagents (e.g., n-Bu₃SnH). These limitations considerably restrict not only the operational simplicity toward the preparation of this

Special Issue: Photocatalysis

Received: April 18, 2016 Published: June 3, 2016

The Journal of Organic Chemistry

pyrroloquinazoline-type tetracyclic compounds but also the pharmaceutical availability of compounds bearing this characteristic scaffold.

Inspired by these beautiful works, as well as our previous work with respect to the photoredox-catalyzed radical triple-bond insertions to construct various heterocycles, 6 we envisioned that the tetracyclic pyrroloquinazolinanes could be prepared by photoredox-catalyzed intramolecular radical cyanide insertion of N-(2-iodobenzyl)-N-acylcyanamides (Scheme 2d). Several challenges associated with this approach have to be pointed out, including (1) the inertia of aryl iodides for the generation of aryl radical species upon photoredox conditions compared with other active aryl radical precursors, such as aryl diazonium salts and diaryliodonium salts, and (2) the inertia of the cyano group as the radical acceptor in photoredox catalysis.

With these considerations in mind, we initially used N-acyl-N-(2-iodobenzyl)-cyanamide (1a)^{4a} as the model substrate to explore this transformation. To our delight, 30% yield of the desired pyrroloquinazoline 2a based on ¹H NMR analysis was obtained when a solution of 1a, TEA and fac-Ir(ppy)₃ (2.0 mol %) in MeCN was irradiated by white LED strips for 12 h at room temperature (Table 1, entry 1). The use of inorganic bases such as Na_2HPO_4 and KO^tBu instead of TEA could not give any of the desired product (entries 2-3). This phenomenon suggests that TEA not only was used as a base but also played as a quencher of photoexcited catalyst (fac-

Table 1. Reaction Condition Optimization^a

entry	photocatalyst	base	solvent	2a (%) ^b
1	Ir(ppy) ₃	TEA	MeCN	30
2	$Ir(ppy)_3$	Na ₂ HPO ₄	MeCN	0
3	$Ir(ppy)_3$	KO^tBu	MeCN	0
4	$Ir(ppy)_3$	Ph_3N	MeCN	0
5	$Ir(ppy)_3$	DIPEA	MeCN	69
6	$Ir(ppy)_3$	DBU	MeCN	trace
7	$Ir(ppy)_3$	DMAP	MeCN	trace
8	$Ir(ppy)_3$	NMM	MeCN	9
9	$Ir(ppy)_3$	TMEDA	MeCN	38
10	$Ir(ppy)_2(dtbbpy)PF_6$	DIPEA	MeCN	66
11	Ir(dFCF ₃ ppy) ₂ (dtbbpy)PF ₆	DIPEA	MeCN	18
12	$Ru(bpy)_3Cl_2$	DIPEA	MeCN	20
13	$Ru(Phen) (PF_6)_2$	DIPEA	MeCN	17
14	$Ir(ppy)_3$	DIPEA	DMF	29
15	$Ir(ppy)_3$	DIPEA	MeOH	0
16	$Ir(ppy)_3Cl_2$	DIPEA	DMSO	64
17	$Ir(ppy)_3$	DIPEA	CH_2Cl_2	35
18 ^c	$Ir(ppy)_3$	DIPEA	MeCN	70
19 ^d	$Ir(ppy)_3$	DIPEA	MeCN	87 (83) ^e
20 ^f	$Ir(ppy)_3$	DIPEA	MeCN	0
21	none	DIPEA	MeCN	0

^aReaction conditions: A solution of **1a** (0.1 mmol), base (0.2 mmol) and photocatalyst (0.002 mmol, 2.0 mol %) in the indicated solvent (1.0 mL) was irradiated by white LED strips at rt for 12 h. ^bThe yields are based on ¹HNMR analysis. ^c0.3 mmol of DIPEA was used. ^d5.0 mL of MeCN was used. ^eIsolated yield. ^fReaction was conducted in the dark.

Ir(ppy)₃*) in the catalytic cycle. We further screened a range of organic bases including Ph₃N, DIPEA, DBU, DMAP, NMM, and TMEDA (entries 4–9). DIPEA was proved to be the optimal organic base affording **2a** in 69% yield (entry 5). The employment of other commonly used photocatalysts (Ir(ppy)₂(dtbbpy)PF₆, Ir(dFCF₃ppy)₂(dtbbpy)PF₆, Ru(bpy)₃Cl₂, and Ru(Phen)₃(PF₆)₂) could not give any improvement of the yield (entries 10–13). Other solvents, such as DMF, MeOH, DMSO, and CH₂Cl₂, were not superior to MeCN in this transformation (entries 14–17). The use of 3.0 equiv of DIPEA gave a comparable yield of **2a** (entry 18). Satisfactorily, the NMR yield of **2a** could be increased to 87% (83% isolated yield) by diluting the reaction system to 0.02 M (entry 19). The reaction could not proceed without photocatalyst or visible light irradiation (entries 20–21).

With the optimized conditions in hand, we then intended to explore the scope of this transformation. In general, the desired cyclization products 2 with a 6/5/6/6 ring system were achieved in acceptable to excellent yields (Scheme 3). The substituents on the aryl of benzoyl motif could be a range of electronic donating or withdrawing functional groups (-OMe, -Me, -F, -Cl, -CF₃, -CN, CO₂Me), affording desired products 2a-2h in 68%-88% yields. The aryl of benzoyl motif could also be 1- or 2-naphthalene (2i, 67% and 2j, 83%) and 1thiophene (2k, 33%). In addition, N-acylcyanamide derived from 2-iodophenylethanamine was also compatible in this transformation to give 21 with a 6/6/6/6 ring system in 30% yield. Cinnamic acid-derived cyanamides were also suitable in this transformation to give tricyclic products 2m-2o in 50%-86% yields, which were not achieved in the AIBN/n-Bu₃SnHmediated radical cascade cyclization.⁴ The reaction could be scaled up to gram scale. When 3 mmol (1.09 g) of 1a was subjected to standard conditions with 1.0 mol % of photocatalyst, the product 2a could be isolated in 85% yield.

To investigate the mechanism of this reaction, a series of emission quenching experiments were performed to acquire further insight into the photoredox catalytic cycle. The experiments revealed that DIPEA quenched the excited state of fac-Ir(ppy)₃ (for details, see the Supporting Information). This result suggests that the reaction might undergo a reductive quenching mechanism. On the basis of the experimental observations, a plausible mechanism is proposed in Scheme 4. The catalytic cycle starts from the photoexcitation of fac-Ir(ppy)₃ upon visible light irradiation to generate the excitedstate photocatalyst (fac-Ir(ppy)₃*). The reductive quenching of fac-Ir(ppy)₃* by DIPEA gives fac-Ir(ppy)₃⁻ and N-centered radical cation species 7.11 Afterward, a single-electron transfer between N-acylcyanamide 1a and fac-Ir(ppy)₃ affords an aryl radical intermediate 3,7 which undergoes intramolecular radical cyclization to give radical intermediate 5.12 Subsequently, oneelectron oxidation of 5 generates aryl cation intermediate 6, following the aromatization to give tetracyclic pyrroloquinazoline 2a.

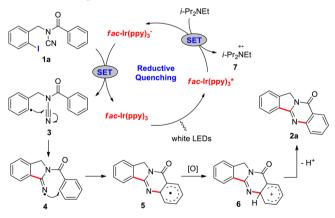
In summary, we have described a practical approach to construct polycyclic pyrroloquinazolines using photoredox-catalyzed intramolecular radical cyanide insertion. This photoredox neutral strategy proceeds under visible light irradiation at room temperature, and the experiments are easy to carry out. The hazardous radical initiator AIBN and toxic *n*-Bu₃SnH can be avoided, which renders this protocol particularly valuable in terms of green synthetic chemistry aspect.

The Journal of Organic Chemistry

Scheme 3. Substrate Scope

"Reaction conditions: A solution of 1a (0.2 mmol), DIPEA (0.4 mmol), and fac-Ir(ppy)₃ (0.004 mmol, 2.0 mol %) in MeCN (10.0 mL) was irradiated by white LED strips at rt for 12 h. Yields are for the isolated products. The reaction was run in 3 mmol scale for 24 h, and the loading of fac-Ir(ppy)₃ was reduced from 2.0 mol % to 1.0 mol %.

Scheme 4. Proposed Mechanism



EXPERIMENTAL SECTION

General Information. All reagents and solvents were used without further purification. Thin-layer chromatography (TLC) was performed on EMD precoated plates (silica gel 60 F254, Art 5715) and visualized by fluorescence quenching under UV light and by staining with phosphomolybdic acid or potassium permanganate, respectively. Column chromatography was performed on EMD Silica Gel 60

(300–400 Mesh) using a forced flow of 0.5–1.0 bar. ¹H NMR (400 MHz), ¹³C NMR (100 MHz), and ¹⁹F (376 MHz) were measured on a 400 M NMR spectrometer. Chemical shifts are expressed in parts per million (ppm) with respect to the residual solvent peak. Coupling constants are reported as Hertz (Hz), signal shapes and splitting patterns are indicated as follows: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet. Infrared (IR) spectra were recorded on a IR spectrophotometer and are reported as wavenumber (cm⁻¹).

General Procedure for Preparation of N-Acyl-N-(2-iodobenzyl)-cyanamides. AB KOH (326 mg, 5.8 mmol) was added to a solution of N-(2-iodobenzyl)-cyanamide (1.5 g, 5.8 mmol) ain THF/ H_2O (15 mL, v:v = 1:1). The mixture was stirred at ambient temperature for 30 min and concentrated under reduced pressure. Toluene (7 mL) was added to the residue. After cooling to 0 °C, acyl chloride (11 mmol) was added slowly. The mixture was then warmed to 25 °C and stirred for another 2 h. After the reaction was complete (as judged by TLC analysis), the reaction mixture was poured into a separatory funnel containing 20 mL of H_2O and 20 mL of CH_2Cl_2 . The layers were separated, and the organic layers were extracted with H_2O (2 × 20 mL). The organic layers were dried over Na_2SO_4 and concentrated under reduced pressure after filtration. The crude product was purified by flash chromatography on silica gel to afford the desired product.

The N-acyl-N-(2-iodobenzyl)-cyanamides derivatives 1a, 1b, 1f, 1g, 1h, and 1k are known compounds and were synthesized according to the literature. ^{4a}

N-Cyano-N-(2-iodobenzyl)-4-methylbenzamide (*1c*). Purification by chromatography (petroleum ether/EtOAc = 15:1) afforded 1c as a white solid (1.7 g, 79% yield). Mp: 84—86 °C; IR (film, cm⁻¹): 2989, 2896, 2230, 1702, 1608, 1567, 1510, 1453, 1435, 835, 756, 742, 725 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.95–7.90 (m, 1H), 7.78 (d, J = 8.2 Hz, 2H), 7.45–7.37 (m, 2H), 7.29 (d, J = 8.0 Hz, 2H), 7.12–7.05 (m, 1H), 4.93 (s, 2H), 2.42 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 168.2, 144.4, 140.2, 135.9, 130.9, 130.7, 129.3, 129.0, 128.8, 127.8, 110.7, 99.3, 55.3, 21.7; HRMS-DART m/z calcd for C₁₆H₁₄ON₂I⁺ [M + H⁺] 377.0145, found 377.0143.

N-Cyano-4-fluoro-N-(2-iodobenzyl)benzamide (*1d*). Purification by chromatography (petroleum ether/EtOAc = 15:1) afforded **1d** as a white solid (1.5 g, 67% yield). Mp: 87—89 °C; IR (film, cm⁻¹): 2989, 2895, 2226, 1707, 1603, 1507, 1440, 844, 752, 745 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.96–7.88 (m, 3H), 7.45–7.37 (m, 2H), 7.21–7.13 (m, 2H), 7.13–7.07 (m, 1H), 4.94 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ167.1, 165.7 (d, J = 256.8 Hz), 140.2, 135.7, 131.7 (d, J = 9.4 Hz), 131.1, 130.9, 128.9, 126.8 (d, J = 3.3 Hz), 116.1 (d, J = 22.3 Hz), 110.4, 99.4, 55.5; ¹⁹F NMR (376 MHz, CDCl₃) δ – 103.55; HRMS-DART m/z calcd for $C_{15}H_{11}ON_2FI^+$ [M + H⁺] 380.9895, found 380.9893.

4-Chloro-N-cyano-N-(2-iodobenzyl)benzamide (1e). Purification by chromatography (petroleum ether/EtOAc = 15:1) afforded 1e as a white solid (1.7 g, 74% yield). Mp: 88—89 °C; IR (film, cm $^{-1}$): 3090, 2975, 2231, 1702, 1590, 1569, 1489, 845, 744 cm $^{-1}$; 1 H NMR (400 MHz, CDCl₃) δ 7.96—7.90 (m, 1H), 7.84—7.79 (m, 2H), 7.49—7.45 (m, 2H), 7.44—7.39 (m, 2H), 7.14—7.06 (m, 1H), 4.93 (s, 2H); 13 C NMR (100 MHz, CDCl₃) δ 167.2, 140.3, 140.0, 135.6, 131.2, 130.9, 130.3, 129.1, 129.0, 128.9, 110.2, 99.4, 55.4; HRMS-DART m/z calcd for $\rm C_{15}H_{11}ON_2ClI^+$ [M + H $^+$] 396.9599, found 396.9599.

N-Cyano-N-(2-iodobenzyl)-1-naphthamide (1i). Purification by chromatography (petroleum ether/EtOAc = 20:1) afforded 1i as a white solid (1.7 g, 72% yield). Mp: 132—134 °C; IR (film, cm⁻¹): 2989, 2898, 2231, 1723, 1698, 1688, 1558, 1467, 801, 779, 765, 742 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.08 (d, J = 8.4 Hz, 1H), 8.03 (d, J = 8.3 Hz, 1H), 7.95 (dd, J = 7.9, 0.8 Hz, 1H), 7.92—7.88 (m, 1H), 7.86 (dd, J = 7.1, 0.8 Hz, 1H), 7.64—7.58 (m, 1H), 7.55—7.55 (m, 1H), 7.55—7.50 (m, 1H), 7.49 (dd, J = 7.7, 1.6 Hz, 1H), 7.42 (td, J = 7.5, 0.9 Hz, 1H), 7.10 (td, J = 7.7, 1.7 Hz, 1H), 5.06 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 168.4, 140.2, 135.8, 133.6, 132.8, 130.84, 130.82, 129.98, 128.9, 128.7, 128.5, 128.0, 127.0, 126.9, 124.4, 124.3, 109.7, 99.4, 54.9; HRMS-DART m/z calcd for $C_{19}H_{14}ON_2I^+$ [M + H⁺] 413.0145, found 413.0141.

N-Cyano-N-(2-iodobenzyl)-2-naphthamide (*1j*). Purification by chromatography (petroleum ether/EtOAc = 20:1) afforded **1j** as a white solid (1.8 g, 75% yield). Mp: 82—84 °C; IR (film, cm⁻¹): 3055, 2986, 2901, 2235, 1703, 1684, 1628, 1558, 1466, 833, 752, 734 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.43 (s, 1H), 7.92 (t, J = 7.4 Hz, 3H), 7.85 (d, J = 8.4 Hz, 2H), 7.62—7.50 (m, 2H), 7.45 (dd, J = 7.6, 1.6 Hz, 1H), 7.39 (t, J = 7.5 Hz, 1H), 7.08 (t, J = 7.6 Hz, 1H), 4.98 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 168.3, 140.2, 135.9, 135.5, 132.1, 131.0, 130.8, 130.4, 129.4, 128.88, 128.89, 128.8, 127.9, 127.8, 127.3, 124.4, 110.6, 99.5, 55.5; HRMS-DART m/z calcd for C₁₉H₁₄ON₂I⁺ [M + H⁺] 413.0145, found: 413.0145.

N-Cyano-N-(2-iodophenethyl)benzamide (*1I*). Purification by chromatography (petroleum ether/EtOAc = 20:1) afforded *1I* as a white solid (1.6 g, 76% yield). IR (film, cm⁻¹): 3085, 3055, 2980, 2230, 1702, 1591, 1568, 1489, 1472, 845, 744 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.86 (d, J = 7.8 Hz, 1H), 7.76–7.70 (m, 2H) 7.61–7.54 (m, 1H), 7.50–7.42 (m, 2H), 7.36–7.28 (m, 2H), 7.70–6.93 (m, 1H), 4.02 (t, J = 7.4 Hz, 2H), 3.24 (t, J = 7.4 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 168.2, 140.0, 139.4, 133.2, 130.9, 130.3, 129.1, 128.7, 128.62, 128.59, 110.9, 100.4, 47.5, 38.6; HRMS-DART m/z calcd for $C_{16}H_{14}ON_2I^+$ [M + H⁺] 377.0145, found: 377.0145.

N-Cyano-N-(2-iodobenzyl)cinnamamide (*1m*). Purification by chromatography (petroleum ether/EtOAc = 40:1) afforded **1m** as a white solid (1.2 g, 53% yield). Mp: 112—114 °C; IR (film, cm⁻¹): 2975, 2930, 2227, 1680, 1617, 1598, 1570, 1468, 745, 748, 687 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.98–7.99 (m, 2H), 7.64–7.58 (m, 2H), 7.47–7.36 (m, 5H), 7.16 (d, J = 15.4 Hz, 1H), 7.10–7.05 (m,

1H), 4.91 (s, 2H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 164.5, 149.1, 140.1, 135.9, 133.5, 131.5, 130.6, 130.2, 129.1, 128.83, 128.79, 113.6, 110.1, 99.2, 54.2; HRMS-DART m/z calcd for $\mathrm{C_{17}H_{14}ON_2I^+}$ [M + H⁺] 389.0145, found: 389.0145.

(E)-N-Cyano-N-(2-iodobenzyl)-3-(p-tolyl)acrylamide (1n). Purification by chromatography (petroleum ether/EtOAc = 40:1) afforded 1n as a white solid (1.4 g, 60% yield). Mp: 129—131 °C; IR (film, cm $^{-1}$): 2968, 2920, 2228, 1695, 1619, 1604, 1568, 1456, 1330, 813, 747, 718 cm $^{-1}$; ¹H NMR (400 MHz, CDCl₃) δ 7.91 (dd, J = 11.6, 3.6 Hz, 2H), 7.50 (d, J = 8.0 Hz, 2H), 7.42—7.32 (m, 2H), 7.23 (d, J = 7.9 Hz, 2H), 7.15—7.02 (m, 2H), 4.91 (s, 2H), 2.40 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 164.7, 149.2, 142.3, 140.1, 136.0, 130.9, 130.6, 130.1, 129.8, 128.9, 128.8, 112.5, 110.2, 99.9, 54.1, 21.6; HRMS-DART m/z calcd for $C_{18}H_{16}ON_2I^+$ [M + H $^+$] 403.0302, found: 403.0302.

(*E*)-*N*-*Cyano*-*N*-(2-iodobenzyl)-3-(4-methoxyphenyl)acrylamide (10). Purification by chromatography (petroleum ether/EtOAc = 20:1) afforded 10 as a white solid (1.6 g, 67% yield). Mp: 135—137 °C; IR (film, cm⁻¹): 2996, 2928, 2228, 1688, 1616, 1599, 1571, 1508, 1458, 1344, 1289, 821, 745 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.93—7.84 (m, 2H), 7.56 (d, J = 8.7 Hz, 2H), 7.43—7.31 (m, 2H), 7.09—7.04 (m, 1H), 7.01 (d, J = 15.3 Hz, 1H), 6.93 (d, J = 8.7 Hz, 2H), 4.89 (s, 2H), 3.85 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 164.8, 162.4, 148.8, 140.1, 136.1, 130.7, 130.5, 130.0, 128.8, 126.3, 114.6, 110.9, 110.3, 99.2, 55.5, 54.1; HRMS-DART m/z calcd for $C_{18}H_{16}O_2N_2I^+$ [M + H⁺] 419.0251, found 419.0251.

General Procedure for the Synthesis of Tetracycle Quinazolinones. A 10 mL round-bottom flask equipped with a rubber septum and magnetic stir bar was charged with *N*-cyano-*N*-(2-iodobenzyl)benzamide 1a (0.2 mmol, 1.0 equiv), DIPEA (0.4 mmol, 2.0 equiv), and Ir(ppy)₃ (0.004 mmol, 0.02 equiv). The flask was evacuated and backfilled with nitrogen 3 times. MeCN (10.0 mL) was added with a syringe under nitrogen. The mixture was then irradiated by white LED strips. After the reaction was complete (as judged by TLC analysis), the mixture was concentrated under reduced pressure after filtration. The crude product was purified by flash chromatography on silica gel (petroleum ether/ethyl acetate = 10:1).

Isoindolo[1,2-b]quinazolin-10(12H)-one (2a).^{4a} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2a as a yellow solid (38.8 mg, 83% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.37 (d, J = 7.8 Hz, 1H), 8.17 (d, J = 7.5 Hz, 1H), 7.86–7.74 (m, 2H), 7.65–7.54 (m, 3H), 7.49 (t, J = 7.4 Hz, 1H), 5.14 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 160.7, 155.0, 149.5, 139.6, 134.3, 132.7, 132.4, 128.9, 127.4, 126.5, 126.4, 123.54, 123.51, 120.6, 49.8.

7-Methoxyisoindolo[1,2-b]quinazolin-10(12H)-one (**2b**). ^{4C} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **2b** as a yellow solid (38.6 mg, 73% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.27 (d, J = 8.8 Hz, 1H), 8.17 (d, J = 7.6 Hz, 1H), 7.66–7.11 (m, 2H), 7.61–7.55 (m, 1H), 7.22 (d, J = 2.4 Hz, 1H), 7.06 (dd, J = 8.8, 2.4 Hz, 1H), 5.13 (s, 2H), 3.95 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 164.6, 160.3, 155.6, 151.8, 139.9, 132.7, 132.3, 128.9, 127.9, 123.5, 123.4, 116.6, 114.2, 108.0, 55.7, 49.7.

7-Methylisoindolo[1,*z*-b]quinazolin-10(12H)-one (2c).^{3b} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2c as a yellow solid (36.7 mg, 74%). ¹H NMR (400 MHz, CDCl₃) δ 8.25 (d, J = 8.1 Hz, 1H), 8.20 (d, J = 7.6 Hz, 1H), 7.66–7.60 (m, 3H), 7.59–7.54 (m, 1H), 7.31 (dd, J = 8.1, 0.9 Hz, 1H), 5.13 (s, 2H), 2.52 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 160.5, 155.1, 149.3, 145.4, 139.7, 132.6, 132.4, 128.9, 128.1, 127.0, 126.3, 123.7, 123.5, 118.1, 49.8, 21.9.

7-Fluoroisoindolo[1,2-b]quinazolin-10(12H)-one (2d). ^{3b} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2d as a yellow solid (37.8 mg, 75%). ¹H NMR (400 MHz, CDCl₃) δ 8.39 (t, J = 7.3 Hz, 1H), 8.18 (d, J = 7.3 Hz, 1H), 7.73–7.56 (m, 3H), 7.48 (d, J = 9.5 Hz, 1H), 7.21 (t, J = 7.7 Hz, 1H), 5.16 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 166.5 (d, J = 253.5 Hz), 160.0, 156.2, 151.7 (d, J = 13.3 Hz), 139.8, 132.7, 132.4, 129.1 (d, J = 11.0 Hz), 129.0, 123.8, 123.6, 117.3, 115.1 (d, J = 23.5 Hz), 112.7 (d, J = 22.2 Hz), 49.9. ¹⁹F NMR (376 MHz, CDCl₃) δ – 103.62;

7-Chloroisoindolo[1,2-b]quinazolin-10(12H)-one (2e). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2e as a yellow solid (47.3 mg, 88%). Mp: 236—238 °C; IR (film, cm⁻¹): 2974, 2920, 1668, 1619, 1597, 1549, 1454, 878, 773 cm⁻¹; 1 H NMR (400 MHz, CDCl₃) δ 8.29 (d, J = 8.5 Hz, 1H), 8.16 (d, J = 7.6 Hz, 1H), 7.81 (d, J = 1.9 Hz, 1H), 7.69—7.56 (m, 3H), 7.43 (dd, J = 8.5, 1.9 Hz, 1H), 5.14 (s, 2H); 13 C NMR (100 MHz, CDCl₃) δ 160.1, 156.1, 150.5, 140.5, 139.8, 132.7, 132.4, 129.0, 127.9, 127.0, 126.9, 123.7, 123.5, 119.0, 49.9; HRMS-DART m/z calcd for $C_{15}H_{10}ON_2Cl^+$ [M + H $^+$] 269.0476, found: 269.0476.

7-(*Trifluoromethyl*) *isoindolo*[1,2-*b*] *quinazolin-10*(12*H*)-*one* (2*f*). ^{3b} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2*f* as a yellow solid (42.3 mg, 70%). ¹H NMR (400 MHz, CDCl₃) δ 8.47 (d, J = 8.3 Hz, 1H), 8.18 (d, J = 7.6 Hz, 1H), 8.11 (s, 1H), 7.74–7.56 (m, 4H), 5.17 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ159.8, 156.2, 149.5, 139.7, 135.8 (q, J = 32.9 Hz), 132.9, 132.2, 129.2, 127.6, 124.9 (q, J = 4.0 Hz), 123.8, 123.6, 123.5 (q, J = 273.7 Hz), 122.8, 122.3 (q, J = 4.1 Hz), 50.0. ¹⁹F NMR (376 MHz, CDCl₃) δ – 63.11.

10-Oxo-10,12-dihydroisoindolo[1,2-b]quinazoline-7-carbonitrile (2g). ^{4a} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2g as a yellow solid (35.3 mg, 68%). ¹H NMR (400 MHz, CDCl₃) δ 8.47 (d, J = 8.1 Hz, 1H), 8.20 (d, J = 7.5 Hz, 1H), 8.15 (s, 1H), 7.72–7.63 (m, 4H), 5.19 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ159.6, 156.6, 149.6, 139.8, 133.2, 132.3, 132.2, 129.3, 128.1, 127.8, 124.0, 123.6, 123.5, 117.9, 117.6, 50.1.

Methyl 10-Oxo-10,12-dihydroisoindolo[1,2-b]quinazoline-7-carboxylate (2h). ^{4a} Purification by chromatography (petroleum ether/ EtOAc = 10:1) afforded 2h as a yellow solid (43.8 mg, 75%). ¹H NMR (400 MHz, CDCl₃) δ 8.51 (s, 1H), 8.43 (d, J = 8.3 Hz, 1H), 8.20 (d, J = 7.5 Hz, 1H), 8.09 (d, J = 8.3 Hz, 1H), 7.72–7.56 (m, 3H), 5.18 (s, 2H), 4.00 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 166.2, 160.2, 155.7, 149.5, 139.7, 135.4, 132.7, 132.5, 129.3, 129.1, 126.8, 126.4, 123.7, 123.6, 123.5, 52.6, 49.9.

Benzo[f]isoindolo[1,2-b]quinazolin-14(12H)-one (2i). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2i as a yellow solid (38.1 mg, 67%). Mp: 247—249 °C; IR (film, cm⁻¹): 2989, 2920, 1656, 1621, 1609, 1592, 1552, 1470, 836, 784, 724 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 9.99 (d, J = 8.6 Hz, 1H), 8.34 (d, J = 8.9 Hz, 1H), 8.16 (d, J = 7.6 Hz, 1H), 8.10 (d, J = 7.7 Hz, 1H), 7.86 (dd, J = 8.2, 5.1 Hz, 2H), 7.83—7.74 (m, 2H), 7.71—7.64 (m, 2H), 5.31 (s, 2H); ¹³C NMR (100 MHz, DMSO) δ 160.6, 156.1, 152.0, 141.4, 135.9, 132.9, 132.7, 131.8, 131.2, 129.2, 129.01, 128.96, 126.9, 126.8, 126.7, 124.7, 123.3, 113.6, 51.1; HRMS-DART m/z calcd for $C_{19}H_{13}ON_2^+$ [M + H⁺] 285.1022, found: 285.1022.

Benzo[h]isoindolo[1,2-b]quinazolin-7(9H)-one (2j). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2j as a yellow solid (47.2 mg, 83%). Mp: 223—225 °C; IR (film, cm⁻¹): 2919, 2850, 1654, 1651, 1610, 1556, 1469, 763, 735, 721 cm⁻¹; 1 H NMR (400 MHz, CDCl₃) δ 9.51–8.94 (m, 1H), 8.33–8.21 (m, 2H), 7.92–7.86 (m, 1H), 7.81 (d, J = 8.7 Hz, 1H), 7.73–7.66 (m, 2H), 7.65–7.53 (m, 3H), 5.15 (s, 2H); 13 C NMR (100 MHz, CDCl₃) δ 160.7, 154.9, 148.4, 139.7, 136.2, 133.1, 132.2, 130.2, 129.1, 128.8, 127.8, 126.7, 126.6, 125.3, 123.6, 123.4, 121.8, 116.8, 50.0; HRMS-DART m/z calcd for $C_{19}H_{13}ON_2^+$ [M + H $^+$] 285.1022, found: 285.1023.

Thieno[3',2':4,5]*pyrimido*[2,1-a]*isoindol-11(9H)-one* (**2k**). ^{4a} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **2k** as a yellow solid (15.8 mg, 33%). ¹H NMR (400 MHz, CDCl₃) δ 8.21 (d, J = 7.6 Hz, 1H), 7.85 (d, J = 5.3 Hz, 1H), 7.67–7.63 (m, 2H), 7.62–7.56 (m, 1H), 7.48 (d, J = 5.3 Hz, 1H), 5.16 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 157.3, 157.0, 156.6, 139.7, 134.6, 132.6, 131.8, 129.1, 124.7, 123.7, 123.6, 121.3, 50.0.

5,6-Dihydro-8H-isoquinolino[1,2-*b*]*quinazolin-8-one* (*2l*). ⁴⁶ Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **2l** as a yellow solid (14.9 mg, 30%). ¹H NMR (400 MHz, CDCl₃) δ 8.54–8.46 (m, 1H), 8.36–8.27 (m, 1H), 7.94–7.72 (m, 2H), 7.55–7.39 (m, 3H), 7.29 (d, J = 7.2 Hz, 1H), 4.42 (t, J = 6.5 Hz, 2H), 3.11 (t, J = 6.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 161.7, 149.5,

147.6, 137.1, 134.3, 131.8, 129.4, 128.1, 127.7, 127.53, 127.50, 126.9, 126.6, 120.7, 39.6, 27.4.

2-Phenyl-2,6-dihydropyrimido[2,1-a]isoindol-4(3H)-one (2m). ¹³ Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2m as a yellow solid (34.1 mg, 65%). ¹H NMR (400 MHz, CDCl₃) δ 7.85 (dd, J = 7.9, 1.1 Hz, 1H), 7.44–7.35 (m, 3H), 7.35–7.27 (m, 2H), 7.26–7.22 (m, 1H), 7.04–6.92 (m, 2H), 4.92 (s, 2H), 4.18 (dd, J = 9.8, 6.0 Hz, 1H), 3.25 (dd, J = 18.2, 9.8 Hz, 1H), 2.84 (dd, J = 18.2, 5.9 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 175.0, 169.6, 139.7, 138.2, 137.3, 129.5, 129.0, 128.3, 128.2, 127.8, 126.4, 97.9, 48.3, 45.1, 38.2.

2-(p-Tolyl)-2,6-dihydropyrimido[2,1-a]isoindol-4(3H)-one (2n). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2n as a yellow solid (47.5 mg, 86%). IR (film, cm $^{-1}$): 3288, 2973, 2920, 1743, 1648, 1514, 1430, 1399, 1346, 818, 733 cm $^{-1}$; 1 H NMR (400 MHz, CDCl₃) δ 7.84 (dd, J = 7.9, 1.0 Hz, 1H), 7.29 (td, J = 7.7, 1.0 Hz, 1H), 7.20 (d, J = 7.9 Hz, 2H), 7.12 (d, J = 8.1 Hz, 2H), 7.03–6.91 (m, 2H), 4.87 (s, 2H), 4.07 (dd, J = 9.8, 6.2 Hz, 1H), 3.20 (dd, J = 18.2, 9.9 Hz, 1H), 2.80 (dd, J = 18.2, 6.2 Hz, 1H), 2.36 (s, 3H); 13 C NMR (100 MHz, CDCl₃) δ 175.1, 147.1, 139.7, 138.1, 137.5, 130.1, 129.9, 129.0, 128.3, 127.7, 126.3, 97.9, 48.2, 44.9, 38.3, 21.1; HRMS-DART m/z calcd for $\rm C_{18}H_{17}ON_2^+$ [M + H $^+$] 277.1335, found: 277.1335.

2-(4-Methoxyphenyl)-2,6-dihydropyrimido[2,1-a]isoindol-4(3H)-one (2o). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 2o as a yellow solid (29.2 mg, 50%). IR (film, cm⁻¹): 3289, 2966, 2918, 1734, 1648, 1609, 1583, 1512, 1430, 1348, 1249, 832, 745 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.85 (d, J = 7.7 Hz, 1H), 7.29 (t, J = 7.3 Hz, 1H), 7.15 (d, J = 8.6 Hz, 2H), 6.98 (t, J = 7.2 Hz, 2H), 6.92 (d, J = 8.7 Hz, 2H), 4.87 (s, 2H), 4.07 (dd, J = 9.9, 6.4 Hz, 1H), 3.82 (s, 3H), 3.20 (dd, J = 18.2, 9.9 Hz, 1H), 2.79 (dd, J = 18.2, 6.4 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ174.9, 170.6, 159.5, 139.7, 137.1, 129.8, 129.1, 129.0, 128.4, 126.6, 114.8, 97.9, 55.4, 48.6, 44.1, 38.2. HRMS-DART m/z calcd for $C_{18}H_{17}O_2N_2^+$ [M + H⁺] 293.1285, found: 293.1284.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b00869.

Characterization of products and copies of ¹H, ¹³C, and ¹⁹F NMR spectra (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: yushouyun@nju.edu.cn.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

Financial support from the National Natural Science Foundation of China (21472084 and 81421091), the Qing Lan Project of Jiangsu Province, and Shanghai Institute of Organic Chemistry. CAS is acknowledged.

■ REFERENCES

(1) (a) Ma, Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. Heterocycles 1997, 46, 541. (b) Ma, Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. Heterocycles 1999, 51, 1883. (c) Ma, Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. Phytochemistry 2000, 53, 1075. (d) Cagir, A.; Jones, S. H.; Gao, R.; Eisenhauer, B. M.; Hecht, S. M. J. Am. Chem. Soc. 2003, 125, 13628. (e) Cheng, K.; Rahier, N. J.; Eisenhauer, B. M.; Thomas, S. J.; Gao, R.; Hecht, S. M. J. Am. Chem. Soc. 2005, 127, 838. (f) Rahier, N. J.; Cheng, K.; Gao, R.; Eisenhauer, B. M.; Hecht, S. M. Org. Lett. 2005, 7, 835. (g) Elban, M. A.; Sun, W.; Eisenhauer, B. M.; Gao, R.; Hecht, S. M. Org. Lett. 2006, 8, 3513.

The Journal of Organic Chemistry

- (2) (a) Thomas, C. J.; Rahier, N. J.; Hecht, S. M. Bioorg. Med. Chem. 2004, 12, 1585. (b) Ma, Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. Bioorg. Med. Chem. Lett. 2004, 14, 1193. (c) Michael, J. P. Nat. Prod. Rep. 2004, 21, 650. (d) Lee, E. S.; Park, J. G.; Kim, S. I.; Jahng, Y. Heterocycles 2006, 68, 151.
- (3) For Pd-catalyzed cyclization to prepare tetracyclic quinazolinones, see: (a) Ju, Y.; Liu, F.; Li, C. Org. Lett. 2009, 11, 3582. (b) Shen, C.; Man, N. Y. T.; Stewart, S.; Wu, X.-F. Org. Biomol. Chem. 2015, 13, 4422.
- (4) For radical approaches to prepare tetracyclic quinazolinones, see: (a) Servais, A.; Azzouz, M.; Lopes, D.; Courillon, C.; Malacria, M. Angew. Chem., Int. Ed. 2007, 46, 576. (b) Bowman, W. R.; Elsegood, M. R. J.; Stein, T.; Weaver, G. W. Org. Biomol. Chem. 2007, 5, 103. (c) Beaume, A.; Courillon, C.; Derat, E.; Malacria, M. Chem. Eur. J. 2008, 14, 1238.
- (5) For reviews on HAS, see: (a) Studer, A.; Curran, D. P. Nat. Chem. **2014**, *6*, 765. (b) Studer, A.; Curran, D. P. Angew. Chem., Int. Ed. **2011**, 50, 5018. (c) Bowman, W. R.; Storey, J. M. D. Chem. Soc. Rev. **2007**, 36, 1803.
- (6) For a review on photoredox-catalyzed radical triple-bond insertion to construct heterocycles, see: (a) Sun, X.; Yu, S. Youji Huaxue 2016, 36, 239. For seminal reports, see: (b) Tong, K.; Zheng, T.; Zhang, Y.; Yu, S. Adv. Synth. Catal. 2015, 357, 3681. (c) Sun, X.; Li, J.; Ni, Y.; Ren, D.; Hu, Z.; Yu, S. Asian J. Org. Chem. 2014, 3, 1317. (d) Cheng, Y.; Yuan, X.; Jiang, H.; Wang, R.; Ma, J.; Zhang, Y.; Yu, S. Adv. Synth. Catal. 2014, 356, 2859. (e) Sun, X.; Yu, S. Org. Lett. 2014, 16, 2938. (f) Jiang, H.; Cheng, Y.; Wang, R.; Zheng, M.; Zhang, Y.; Yu, S. Angew. Chem., Int. Ed. 2013, 52, 13289. (g) Cheng, Y.; Jiang, H.; Zhang, Y.; Yu, S. Org. Lett. 2013, 15, 5520.
- (7) For examples on photoredox activation of iodides, see: (a) Ghosh, I.; Ghosh, T.; Bardagi, J. I.; König, B. Science 2014, 346, 725. (b) Cheng, Y.; Gu, X.; Li, P. Org. Lett. 2013, 15, 2664. (c) Nguyen, J. D.; D'Amato, E. M.; Narayanam, J. M. R.; Stephenson, C. R. J. Nat. Chem. 2012, 4, 854. (d) Kim, H.; Lee, C. Angew. Chem., Int. Ed. 2012, 51, 12303.
- (8) For examples on photoredox activation of diazonium salts, see: (a) Wang, H.; Yu, S. Org. Lett. 2015, 17, 4272. (b) Hari, D. P.; König, B. Angew. Chem., Int. Ed. 2013, 52, 4734. (c) Hari, D. P.; Schroll, P.; König, B. J. Am. Chem. Soc. 2012, 134, 2958. (d) Pratsch, G.; Anger, C. A.; Ritter, K.; Heinrich, M. R. Chem. Eur. J. 2011, 17, 4104. (e) Kalyani, D.; McMurtrey, K. B.; Neufeldt, S. R.; Sanford, M. S. J. Am. Chem. Soc. 2011, 133, 18566. (f) Wetzel, A.; Pratsch, G.; Kolb, R.; Heinrich, M. R. Chem. Eur. J. 2010, 16, 2547. (g) Wetzel, A.; Ehrhardt, V.; Heinrich, M. R. Angew. Chem., Int. Ed. 2008, 47, 9130. (h) Heinrich, M. R.; Wetzel, A.; Kirschstein, M. Org. Lett. 2007, 9, 3833.
- (9) For examples on photoredox activation of diaryliodonium salts, see: (a) Jiang, H.; Cheng, Y.; Wang, R.; Zhang, Y.; Yu, S. Chem. Commun. 2014, 50, 6164. (b) Fumagalli, G.; Boyd, S.; Greaney, M. F. Org. Lett. 2013, 15, 4398. (c) Neufeldt, S. R.; Sanford, M. S. Adv. Synth. Catal. 2012, 354, 3517.
- (10) To the best of our knowledge, there is only one example on the radical cyanide insertion using photoredox catalysis, see ref 6c.
- (11) (a) Nicewicz, D. A.; MacMillan, D. W. C. Science 2008, 322, 77. (b) Ischay, M. A.; Anzovino, M. E.; Du, J.; Yoon, T. P. J. Am. Chem. Soc. 2008, 130, 12886. (c) Narayanam, J. M. R.; Tucker, J. W.; Stephenson, C. R. J. J. Am. Chem. Soc. 2009, 131, 8756. (d) Furst, L.; Matsuura, B. S.; Narayanam, J. M. R.; Tucker, J. W.; Stephenson, C. R. J. Org. Lett. 2010, 12, 3104.
- (12) For intramolecular radical cyclization of cyanides, see:
 (a) Vervisch, K.; D'hooghe, M.; Törnroos, K. W.; Kimpe, N. D. Org. Biomol. Chem. 2012, 10, 3308. (b) Larraufie, M.-H.; Ollivier, C.; Fensterbank, L.; Malacria, M.; Lacôte, E. Angew. Chem., Int. Ed. 2010, 49, 2178. (c) Fernandez-Mateos, A.; Teijón, P. H.; Burón, L. M.; Clemente, R. R.; González, R. R. Synlett 2007, 72, 9973. (d) Benati, L.; Bencivenni, G.; Leardini, R.; Minozzi, M.; Nanni, D.; Scialpi, R.; Spagnolo, P.; Zanardi, G.; Rizzoli, C. Org. Lett. 2004, 6, 417. (e) Camaggi, C. M.; Leardini, R.; Nanni, D.; Zanardi, G. Tetrahedron 1998, 54, 5587. (f) Curran, D. P.; Liu, H.; Josien, H.; Ko, S.-B.

Tetrahedron 1996, 52, 11385. (g) Nanni, D.; Pareschi, P.; Rizzoli, C.; Sgarabotto, P.; Tundo, A. Tetrahedron 1995, 51, 9045. (13) Liu, F.; Li, C. J. Org. Chem. 2009, 74, 5699.